Novel Production of 3-Benzoyl-2,1-benzisoxazoles from 2-Phenyl-quinolin-4(1H)-ones

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2-Phenylquinolin-4(1*H*)-ones **1a** and **1b** react with sodium dichloroisocyanurate **2** in methanolic aqueous sodium hydroxide, to afford, after acidification, 3-benzoyl-2,1-benzisoxazoles **3a** and **3b** (*ca.* 30%).

Two basic routes are available for the synthesis of 2,1-benzisoxazoles (anthranils), the subject of many investigations and two reviews: $^{1,2}(i)$ cyclization of types (a) and (b), in which the 1–2 and 2–3 bond, respectively, is formed, and (ii) introduction of C-3, forming bonds 2–3 and 3–3a (c). There are also a number of less general and mostly multi-step synthetic methods. 1,2

We report here a novel, mild and essentially one-pot synthesis of 3-benzoyl-2,1-benzisoxazole 3a and its 5-methyl derivative 3b, via a remarkable transformation of the corresponding 2-phenyl-quinolin-4(1H)-ones 1a and 1b. Thus, treatment of the readily available 1,3 in methanolic aqueous

Scheme 1 Reagents and conditions: i, 1 (4 mmol) in MeOH–2 mol dm $^{-3}$ NaOH–H₂O (2:4:1) (90 ml), stir, sodium dichloroisocyanurate 2 (9 mmol) added in one portion, room temperature, 1 h; ii, chill, then conc. HCl (11 ml), continued stirring, 15 min; filter; iii, sparingly soluble 3 (plus other material) with MeOH–2 mol dm $^{-3}$ NaOH–H₂O (1:1:1) (45 ml), stir, 15 min, filter, crystallization (Me₃OH–H₂O)

sodium hydroxide, with excess of sodium dichloroisocyanurate **2**,⁴ and subsequent acidification of the reaction, afforded the aforementioned **3** (*ca.* 30%, not optimized). The constitution of product **3a**, C₁₄H₉NO₂, from **1a** was proved by direct comparison [IR (KBr), ¹H NMR (CDCl₃), mixed m.p.] with authentic 3-benzoyl-2,1-benzisoxazole **3a** prepared^{5,6} from 2-phenylisatogen. The identity of the analogous product, m.p. 114–115 °C, from **1b** (previously⁴ incorrectly formulated as 6-methyl-2-phenyl-4*H*-3,1-benzoxazin-4-one), was unequivocally established as the new 3-benzoyl-5-methyl-2,1-benzisoxazole **3b** from an X-ray crystal structure determination (Fig. 1).† The bond lengths and angles in **3b** are virtually the same as the corresponding ones in **3a**.⁷

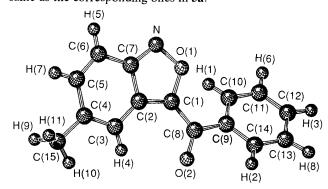


Fig. 1 The molecular structure and atomic numbering scheme for compound $3\ensuremath{b}$

† Crystal data: compound 3b, $C_{15}H_{11}NO_2$, $M_r=237.26$, monoclinic, space group C2/c, a=24.415(4), b=5.753(1), c=17.143(2) Å, $\beta=94.483(7)^\circ$, V=2400.9(7) ų, Z=8, $D_c=1.313$ g cm⁻³, F(000)=992, $\mu=0.51$ cm⁻¹, R=0.063 for 2560 unique reflections with $F_o\geq 3\sigma(F_o)$, $R_w=0.059$ where $w \propto 1/\sigma^2(F)$. Data were collected using an Enraf-Nonius CAD4 diffractometer with Mo-K α radiation ($\lambda=0.71069$ Å). The structure was solved by direct methods and refined by full-matrix least-squares analysis with anisotropic thermal parameters for all nonhydrogen atoms. Hydrogen atoms were refined isotropic ally. Lorentz–polarisation and empirical absorption corrections were applied to the data. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

The tormation of 3 from 1 is the result of a series of reactions and intermediates that includes chlorination of 1 to a 3-chloroquinoline and subsequently to the corresponding, hitherto inaccessible, 3,3-dichloro-quinolin-4(3H)-one, and eventual loss of C-3 possibly via a decarboxylative elimination.8 Studies are continuing to probe the scope and mechanistic aspects of this rapid and convenient method for accessing the 2,1-benzisoxazole ring system from a quinolinone derivative; to date only the reverse situation has been documented.1,2

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References

- 1 R. K. Smalley, Adv. Heterocycl. Chem., 1981, 29, 34.
- 2 K. H. Wünsch and A. J. Boulton, Adv. Heterocycl. Chem., 1967, 8, 303.
- 3 B. Staskun and S. S. Israelstam, J. Org. Chem., 1961, 26, 3191.
 4 B. Staskun, J. Org. Chem., 1988, 53, 5287.
 5 F. Kröhnke and M. Meyer-Delius, Berichte, 1951, 84, 932.

- 6 J. L. Pinkus, G. G. Woodyard and T. Cohen, J. Org. Chem., 1965, 30, 1104.
- 7 M. Sundaralingam and G. A. Jeffrey, Acta Crystallogr., 1962, 15,
- 8 B. Staskun and T. van Es, unpublished results.